Evaluating Ammonium, Nitrate and Sulfate Aerosols in 3-Dimensions

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MATRIX EQSAM

OMA EQSAM

WUSA

MATRIX ISORROPIA II

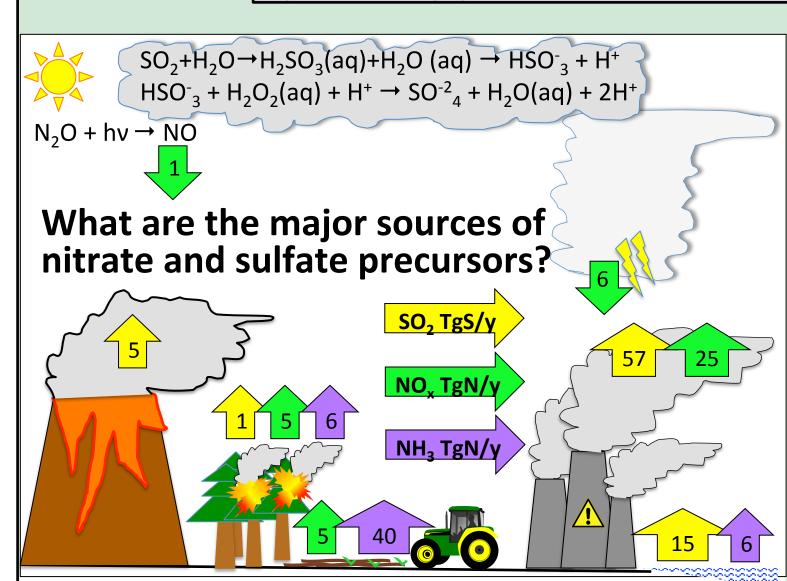
Background & Motivation

The effect aerosols have on climate and air quality is a function of their chemical composition, concentration and spatial distribution. These parameters are controlled by emissions, heterogeneous and homogeneous chemistry, where thermodynamics plays a key role, transport, which includes stratospherictropospheric exchange, and depositional sinks. In this work we demonstrate the effect of some of these processes on the SO_4 -NH₄-NO₃ system using the GISS ModelE2 Global Circulation Model (GCM).

- Motivation: NO₃ aerosol is poorly constrained throughout the troposphere, especially above surface level.
- Mission: Bridge this knowledge gap with a collection of surface and airborne data and aerosol models. Relevant studies: Bauer at al., 2007, Bellouin et al., 2011, Aan de Brugh et al., 2012, Hauglustaine et al., 2014,

Objective: evaluate the GISS ModelE2 aerosol schemes and pin point key process either included or missing in the model

L. GAS PHASE



How does NO₃ aerosol 2. Dissolve in water from? 3. Dissociate into ions

 $2H_2O(aq) \rightleftharpoons OH^-(aq) + H_3O^+(aq)$ H_2O $H_2SO_4(g) \neq H_2O(aq) \rightleftharpoons HSO_4(aq) + H_3O(aq)$ $SO_2 \rightarrow H_2SO_4$ $HNO_3(g) + H_2O(aq) \rightleftharpoons NO_3^-(aq) + H_3O^+(aq)$ $NH_3(g) + H_2O(aq) \rightleftharpoons NH_4^+(aq) + OH^-(aq)$ $NO_x \rightarrow HNO_3$

NH₄NO₂ form? The phase is controlled by: L. Precursor abundance

4. Ionic solution – salt equilibrium $HSO_4^{-1}(aq) + H_2O(aq) \rightleftharpoons SO_4^{2-1}(aq) + H_3O^{+1}(aq)$ $NH_4^+(aq) + HSO_4^-(aq) \rightleftharpoons NH_4HSO_4$ (s) $2NH_4^+(aq) + SO_4^{2-}(aq) \rightleftharpoons (NH_4)_2SO_4^-(s)$ $NH_4^+(aq) + NO_3^-(aq) \rightleftharpoons NH_4NO_3 (s)$

Ambient Relative humidity (RH)

Figure 2: The processes controlling NO₃ formation

Setup:

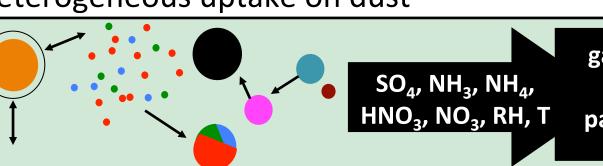
- 2° x 2.5° resolution, 40 vertical layers
- Fully interactive trop and strat chemistry
- Horizontal wind nudged: 6-hourly NCEP
- SST and Ice Cover prescribed from obs
- **Emissions:**
- CMIP5
- RCP4.5 (2005 onwards)

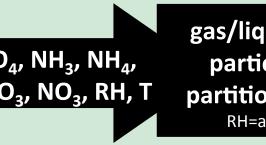
Paulot et al., 2015

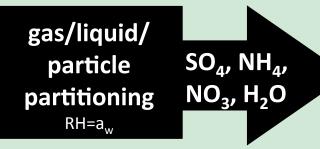
- Biomass burning: GFED3
- Agricultural NH₃ imposed seasonality according to solar zenith angle

GISS Aerosol Schemes

- MATRIX: microphysics model, tracks mixing state
- **OMA**: bulk aerosol, includes heterogeneous uptake on dust
- **EQSAM**: parameterized thermodynamics **ISORROPIA II**: calculates the thermodynamics







Sensitivity runs: (Result 3) we test the sensitivity of NH₃, NH₄, HNO₃, NO₃ to doubling and fivefold increase in agricultural NH₃ emissions.

Data and Methods

Monthly mean Surface data of SO₄, NH₃, NH₄, NO₃ measured via the Western USA (WUSA) [30°-50°N, 114°-130°W] IMPROVE (USA) and EMEP-NILU (EU) networks during 2000-2010 is used to compare against the simulations. From the climatological mean (Figure 4) we adopt a regional approach (black frames in Figures 3,4), where the mean, standard deviation, normalized mean bias and correlation coefficients are calculated for the stations within a region along with their matching model grid boxes.

During 2001-2011, 14 flight campaigns took place in the NH and measured SO₄, NH₄, HNO₃, NO₃ (Figure 3). The flights were predominantly during spring and summer time and deployed the AMS instrument. With a regional approach we parse out transit flights and for flights within the ARC, EUSA, WUSA regions we use the data within the regional boundaries to calculate a campaign mean per model layer. We sample our simulations according to the flight location.

[55°-90°N, 60°-170°W] **Arctic (ARC)** [30°-50°N, 60°-95°W] Eastern USA (EUSA) |European Union (EU) [35°-70°N, 10°W-30°E]

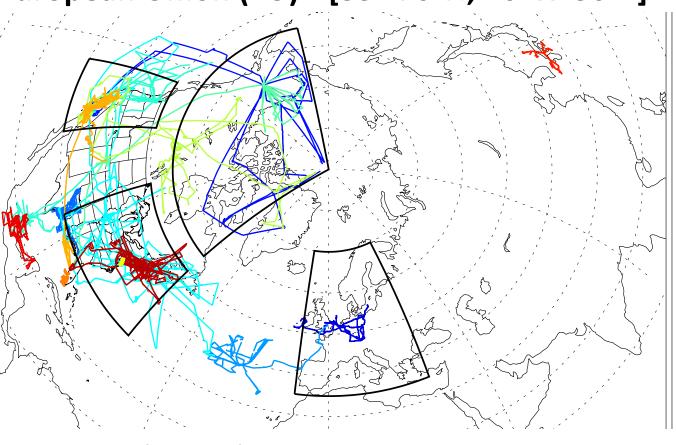


Figure 3: (above) Flight tracks of 14 flight campaigns used in this study

References

Bauer and Koch (2005), J. Geophys. Res. Bauer et al. (2007a), J. Geophys. Res. Bauer et al. (2007b), Atmos. Chem. Phys. Bauer et al. (2008), Atmos. Chem. Phys. Fountoukis and Nenes (2007), Atmos. Chem. Phys. Fowler et al. (2015), Atmos. Chem. Phys. Koch et al. (2006), J. Geophys. Res. Atmos.

Koch et al. (2007), J. Geophys. Res. Atmos. Lamarque et al. (2013), Atmos. Chem. Phys. Lamarque et al. (2010), Atmos. Chem. Phys. Metzger et al. (2002a), J. Geophys. Res. Atmos. Metzger et al. (2002b), J. Geophys. Res. Atmos. Schmidt et al. (2014), J. Adv. Model. Earth Syst. Shindell et al. (2001), J. Geophys. Res. Shindell et al. (2006), Atmos. Chem. Phys.

Results Figure 5: (below) Surface regional statistics (2000-2010); correlation coefficient (R)

0-0.3

5 10 15
ARCTAS spring (ARC)

MATRIX SORROPIA

INTEX-A (EUSA)

CALNEX (WUSA)

HNO.

HNO

INTEX-A (EUSA)

Figure 7: (above)

profile correspo-

nding to ARCTAS

spring, INTEX-A,

Mean regional

Is there a spatial pattern (Figure and normalized mean bias (NMB) for the three simulations. 4)? Surface concentrations show high concentrations in EUSA, EU and low concentrations in WUSA. The statistics shows (Figure 5):

- Performance is controlled by region more than aerosol
- Systematic underestimation of aerosols in EUSA, EU
- Big differences for SO₄ with microphysics (MATRIX VS OMA)
- Overall good performance by the GCM (R>0.5)

NO₃ annual cycle (Figure 6): Seasonality is reproduced in

one standard

days sampled

5 10 15

INTEX-A (EUSA)

SO₂₁

 SO_4

SO

INTEX-A (EUSA)

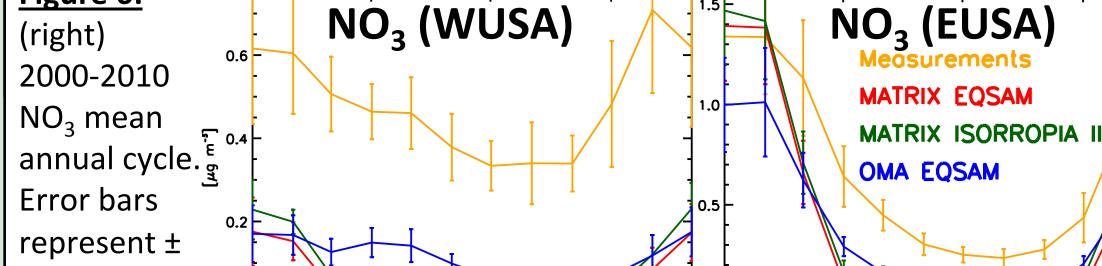
 $\mathsf{NH}_{\mathsf{4}}|_{\mathsf{6}}$

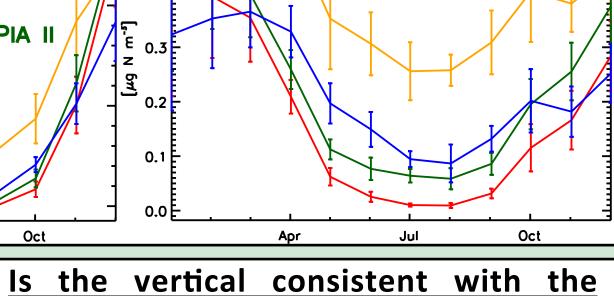
 $|\mathbf{NH_4}|_6$

deviation.

EUSA, EU Summer underestimations in all regions



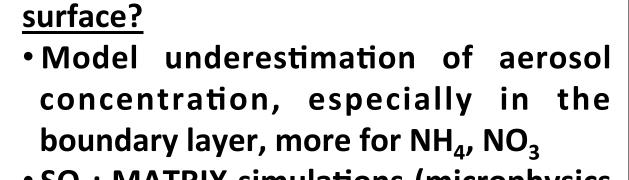




 NO_3 (EU)

0.5

0.4



• SO₄: MATRIX simulations (microphysics included) show higher concentrations than the bulk scheme (OMA). Thermodynamic scheme makes a minor difference: green and red lines overlaying each other.

 The model overestimates upper tropospheric HNO₃: observed and attributed in previous studies to overestimation of strat-trop exchange. Missing processes? (1) Heterogeneous chemistry sink on dust surfaces included in OMA, yields lower concentrations. (2) Homogenous chemistry: (i) Crustal (Mg, K, Li) and sea salt (Na, Cl) ions are not part of the thermodynamics implemented in the model. (ii) Thermodynamics do not take into account the different time scales associated with particle size distribution. (iii) Organic nitrate formation, especially important in the



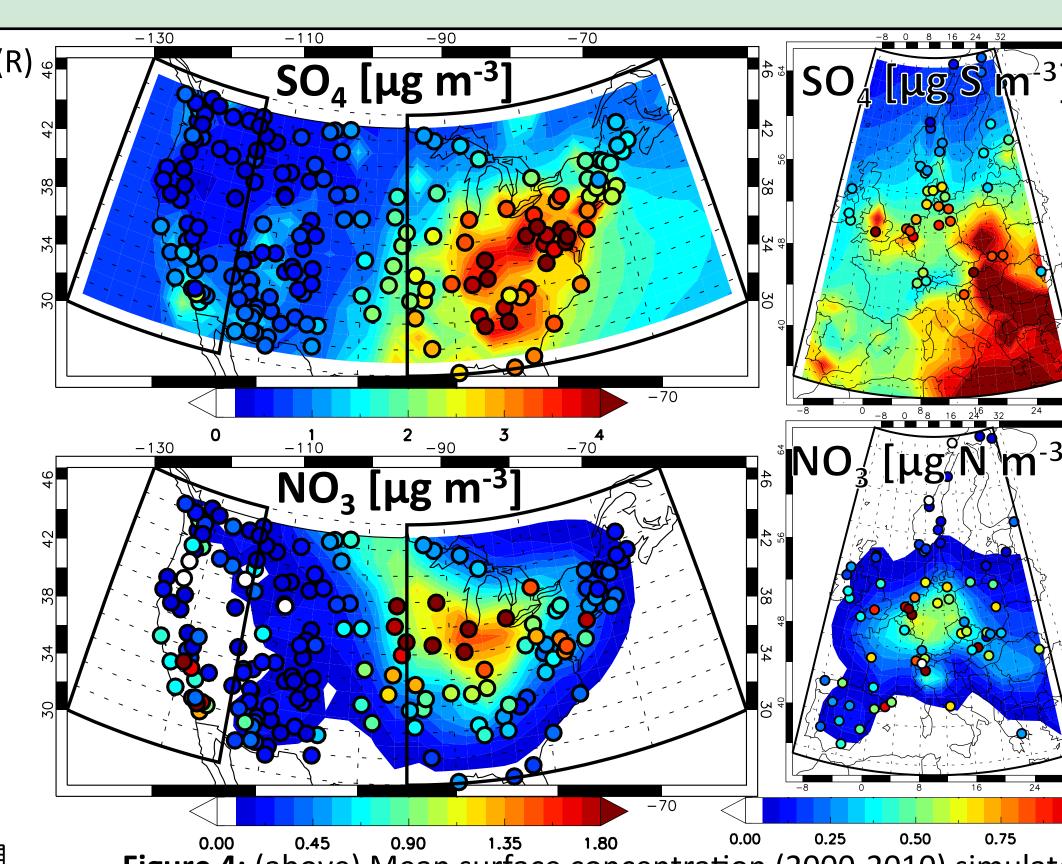
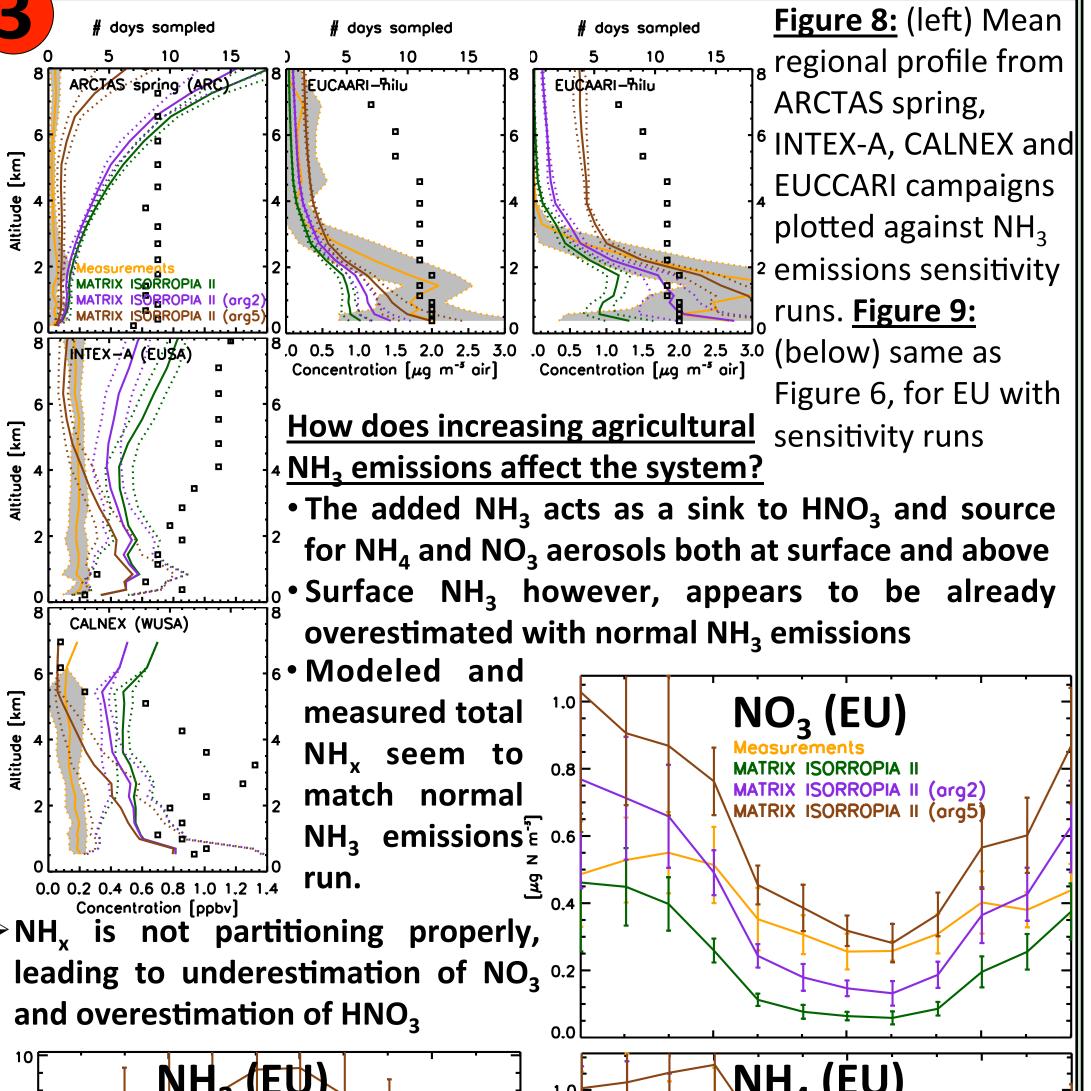
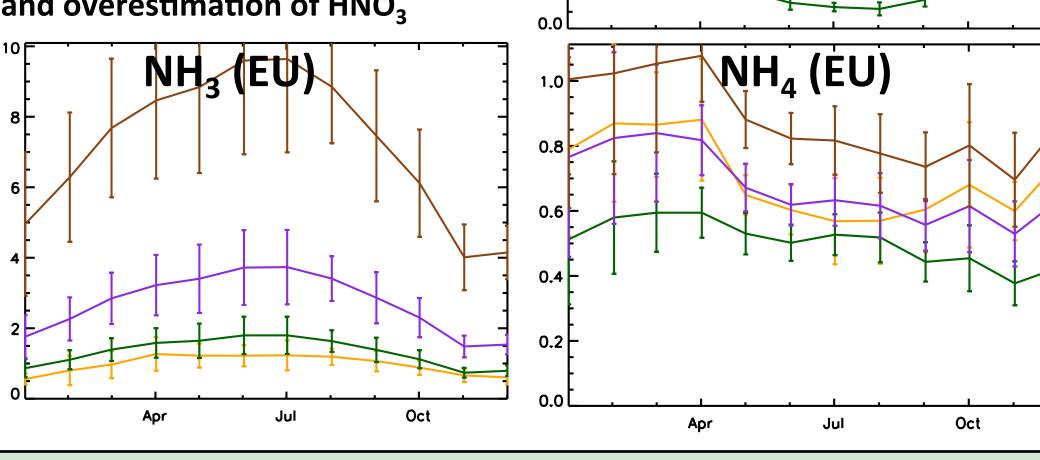


Figure 4: (above) Mean surface concentration (2000-2010) simulated by MATRIX-EQSAM overlaid by measurements





Take away:

- > Missing aerosol mass, especially above the surface, could have important implications to aerosol radiative forcing
- > Good correlations (R>0.5) at surface in regions where seasonality is reproduced
- \triangleright HNO₃ is sensitive to the heterogeneous sink an important process to include in models
- > HNO₃ and NO₃ partitioning is strongly dependent on NH_x partitioning
- \triangleright Need for more measurements: few campaigns measured NH₃ (CALNEX, TexAQ), no winter campaigns